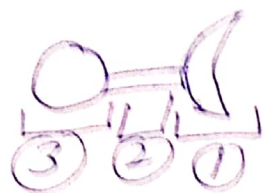


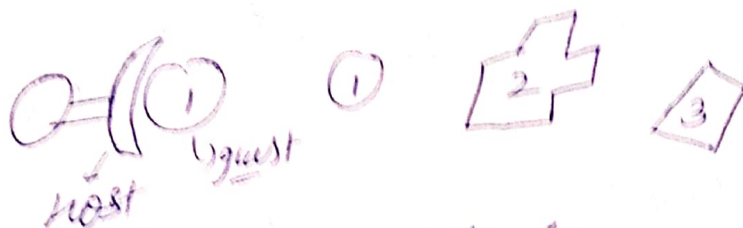
Supramolecular chemistry

① Supramolecular Assembly.

② Molecular receptor / Molecular substrate:-



- ① Receptor
- ② Spacer
- ③ Signal



③ Molecular recognition:-

Anionic recognition

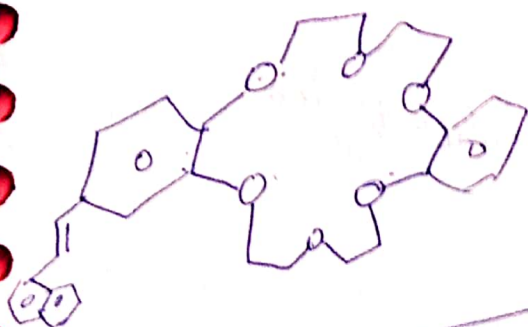
Cationic recognition

molecular recognition

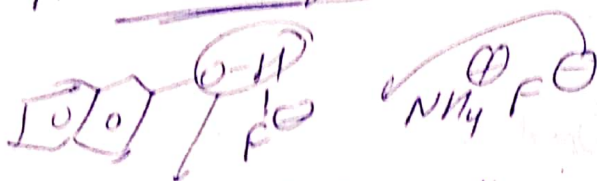
+ Cation recognition:-
Crown-ether



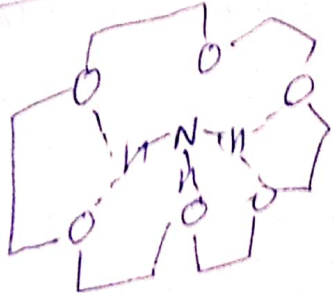
Cationic receptor
18-crown-6
18-crown-6



Anion recognition

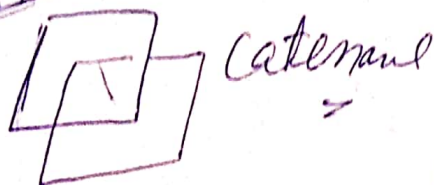
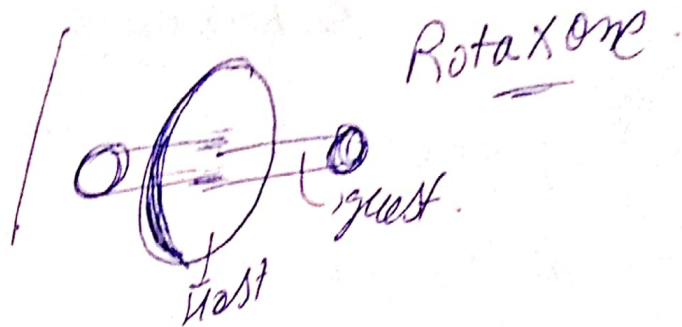
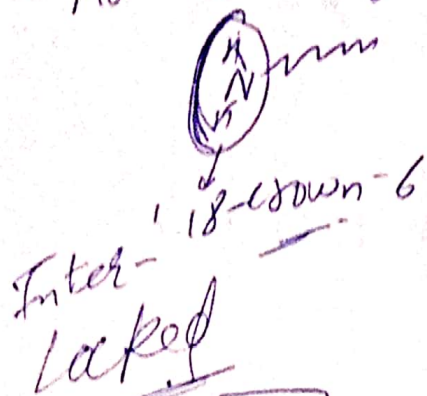


Anion recognition
anion receptor

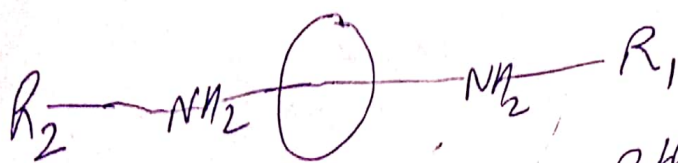


H-bonding molecular recognition

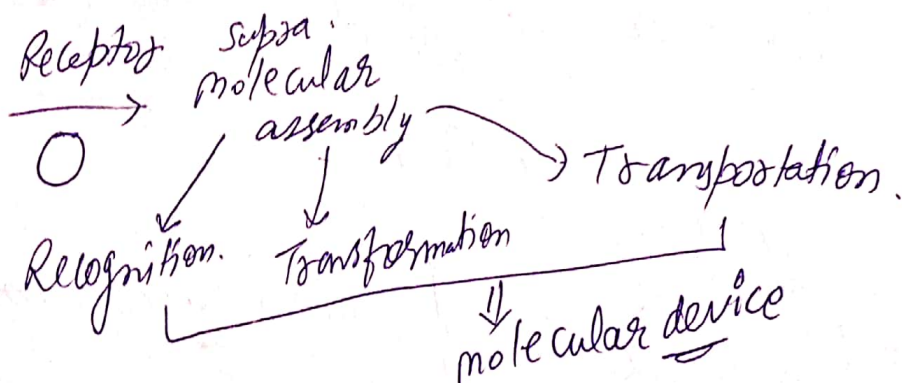
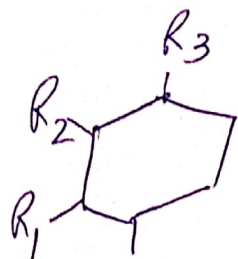
Molecular recognition



#1 molecular machine:-



with change of pH This ring can move to and fro.

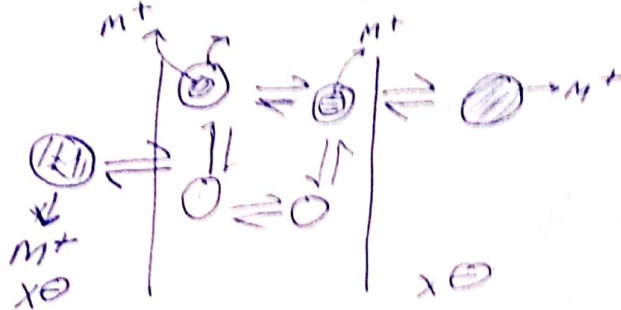


Applications

- ① materials technology. (molecular $\rightarrow$ supramolecular $\rightarrow$ Material)
- ② supramolecular catalysts
- ③ medicine $\rightarrow$ Drug delivery

- ④ Green chemistry ⑤ High-Tech devices
solar cells, LED etc

Transport Process and Carrier design.



To design the

① Transport effect.

② To device the transport process

③ To Investigate the application in chemistry and Biology.

* Carrier mediated transport process →

① Formation of carrier substrate complex in one interface.

② Diffusion of the complex through the membrane

③ Release the substrate at the other interface.

④ Back diffusion of carrier.

* Selective Transport

Internal
from carrier

External
comes from medium.

* Rate depends on the thermodynamic eq^m of both the interfaces.

Transport :-

- Carrier → ① High Selectivity
- ② High stability.
- ③ slow Diffusion and exchange Rate.

Can Transport

- ① Cation
- ② Anion
- ③ Molecule.

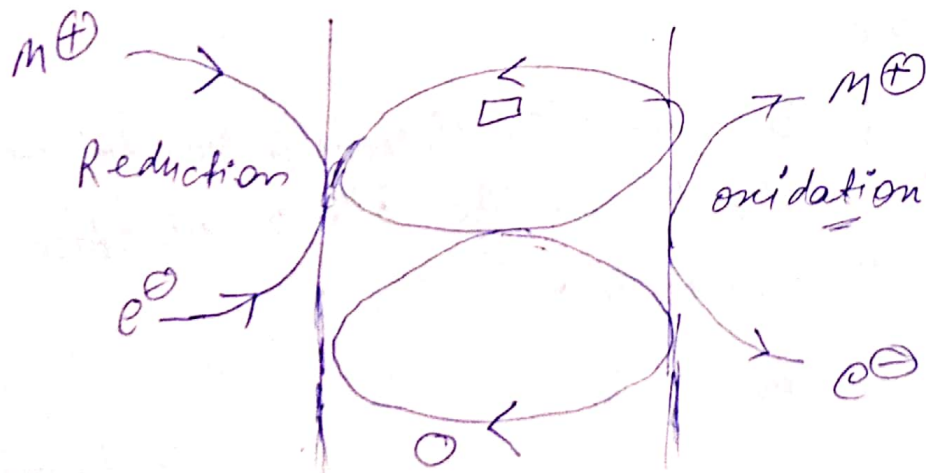
↓
Crown ether
etc
macrocyclic
macrobicyclic
Compounds
↓
Carriers

↓
 NH_4^+ or R-NH_3^+
↓
Carriers

different reducti
on
Pot.
so
Redox
gra

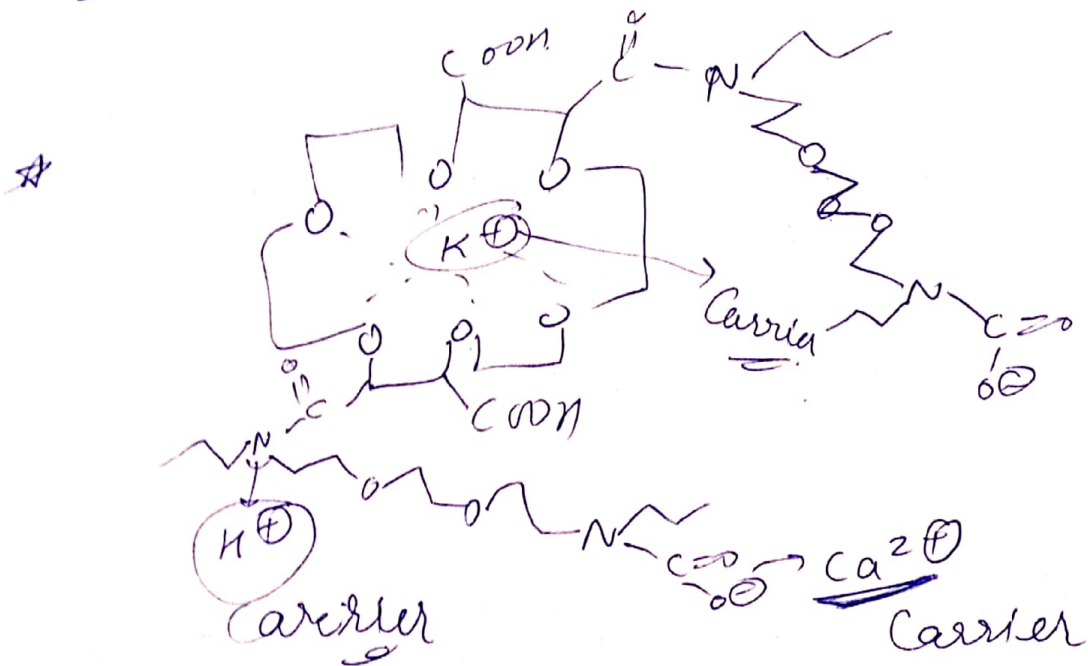
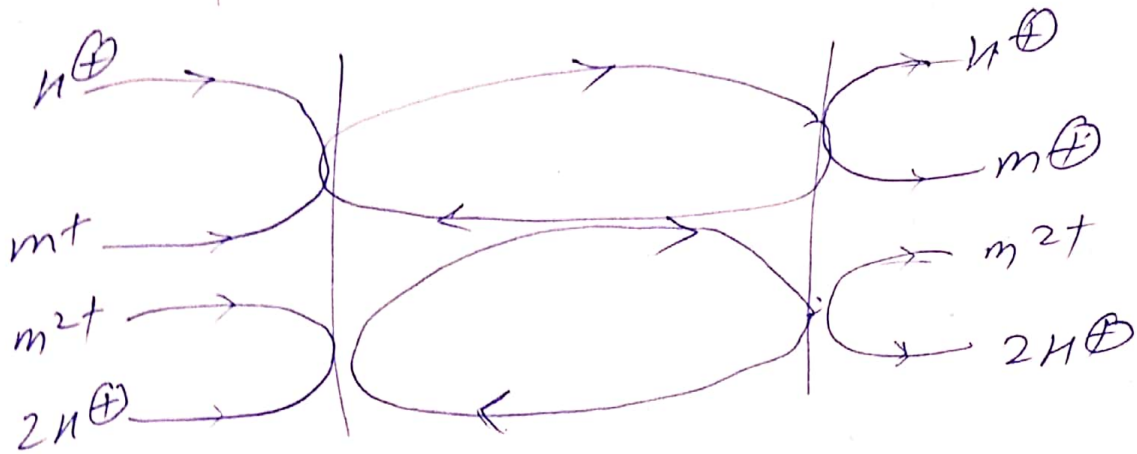
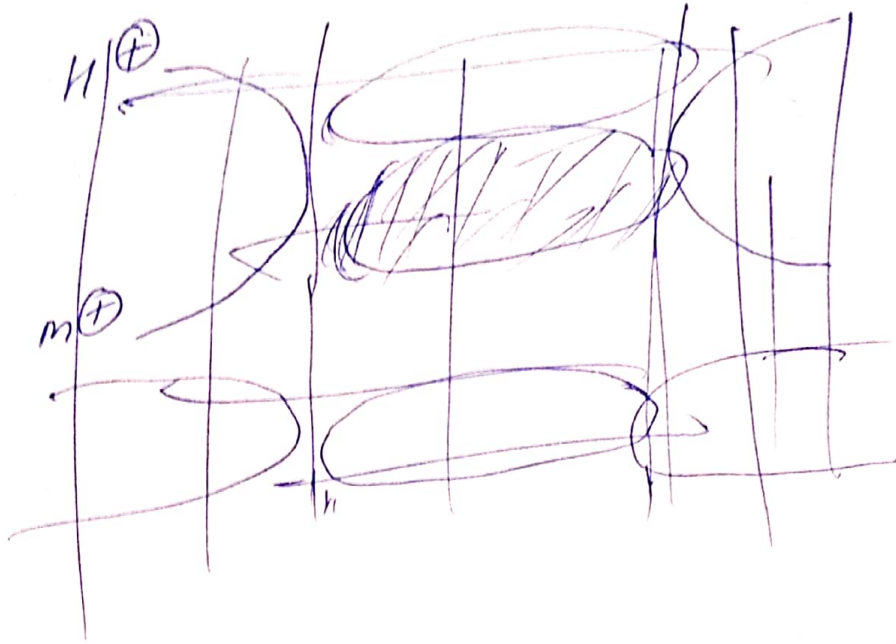
Coupled Transport Process ⇒

- ① Electron coupled transport in redox gradient.

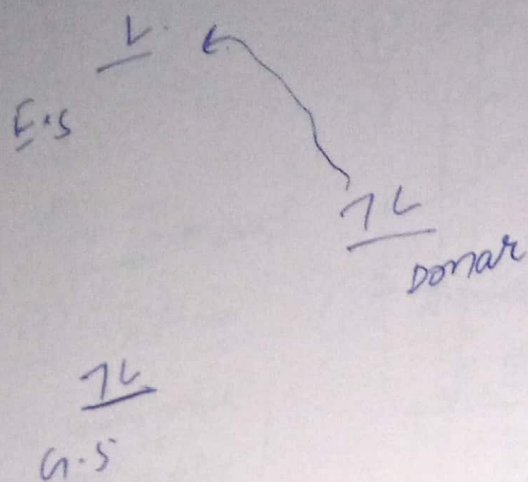


□ → metal ion carrier.
→ macrocycle Ligand en → 18-crown-6
500 K +
○ → c1ccc(cc1)S(=S)(c2ccc(cc2)S(=S)N(c3ccc(cc3)S(=S)c4ccc(cc4)S(=S)c5ccccc5)c6ccccc6)c7ccccc7 - do e^- transport

② Proton coupled transport Process $\rightarrow$ act against pH gradient



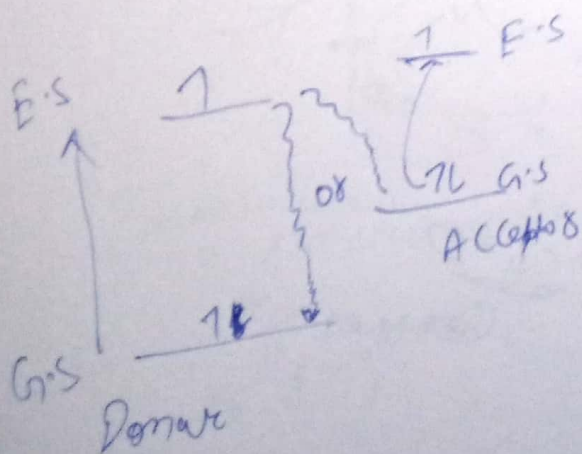
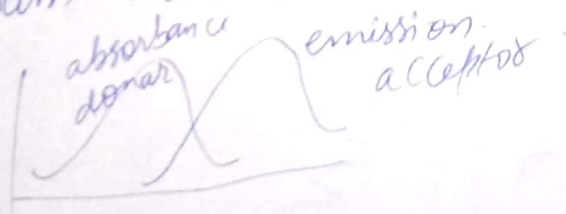
e^- transfer

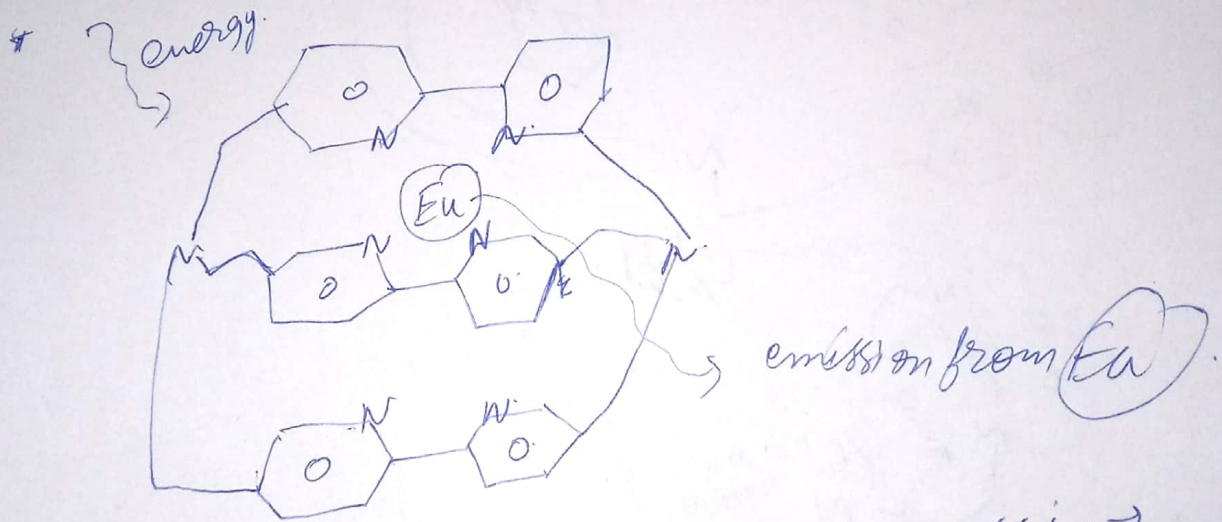
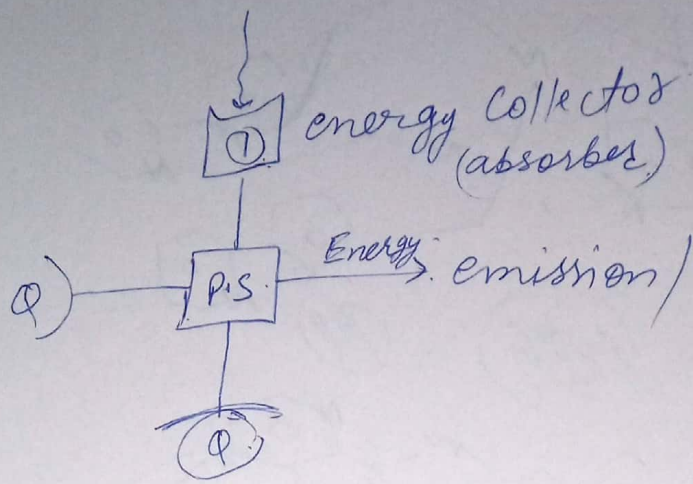


Energy transfer →

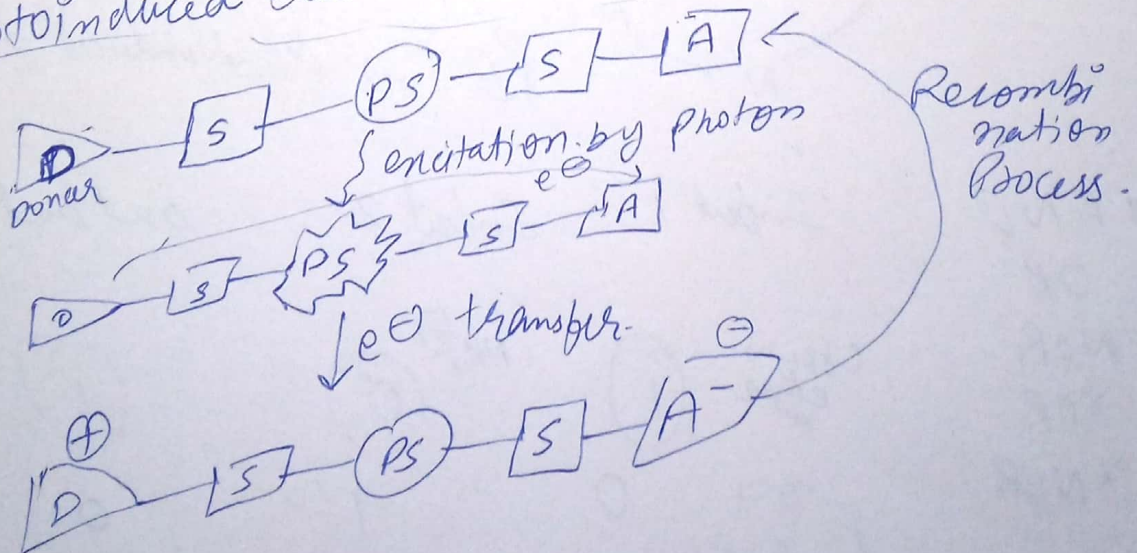
- ① Dexter
- ② Föster

* Minimum overlap

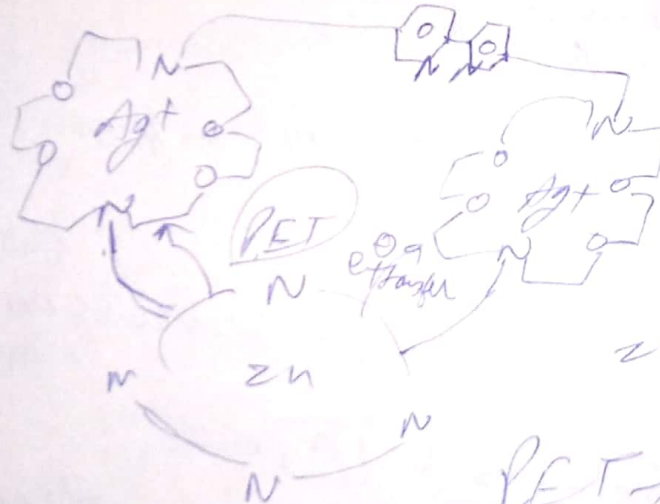
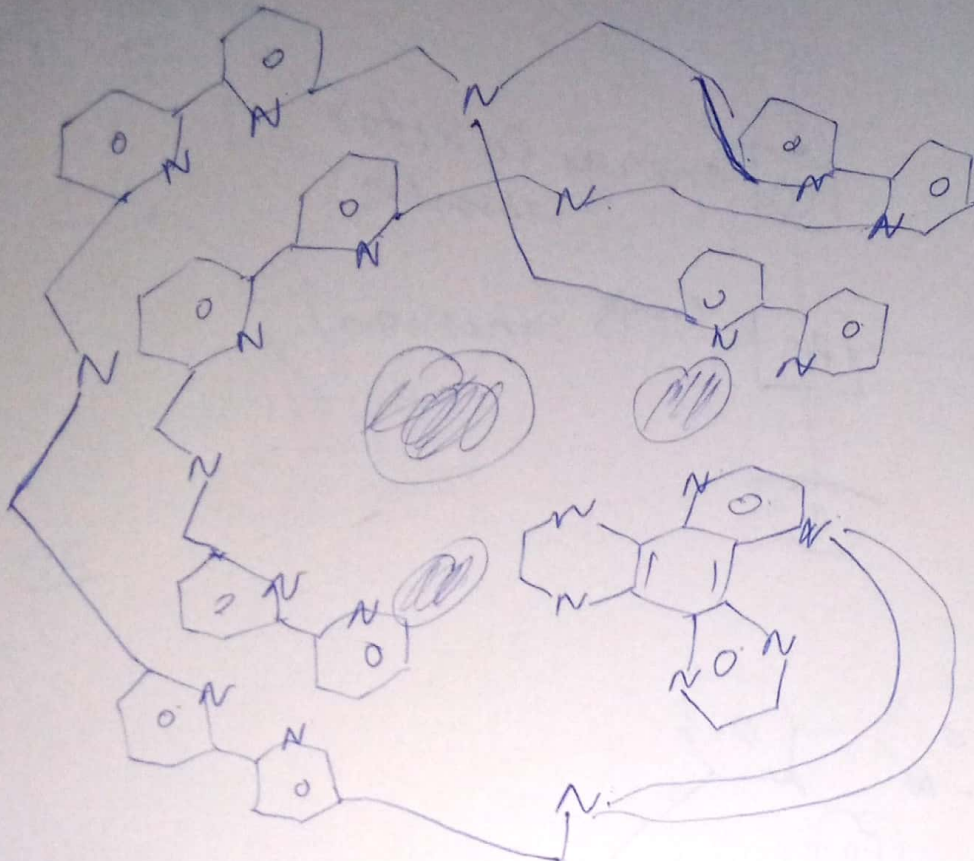




Photoinduced electron transfer process in →



examples → Bi-pyridyl ligands.



Zn-porphyrin

PET $\rightarrow$ Photoinduced e^- transfer

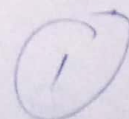
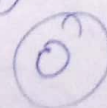
- # AND
- # OR
- # NOR
- # XOR
- # XNOR

Input 1 Input 2 output

Crown ether



NH_4^+



0

1

0

1

1

1 or 0

