

(a) 7

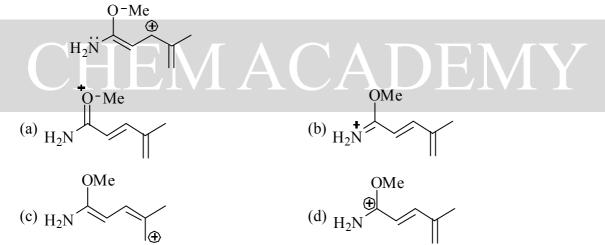
ORGANIC CHEMISTRY

Daily Assignment IIT-JAM Home Work DPP

GOC (2)

1. Number of atoms having sp² hybridisation in the given compounds

2. The most stable resonating structure for the given molecule



3. Among the following which is not represent pair of resonating structure

$$(a) \bigcup_{\Theta} O \bigoplus_{N \in \mathbb{N}} H \bigoplus_{N \in \mathbb{N}} O \bigoplus_{N \in$$

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4. In correct statement about the following molecule.

(a) It contain 4 sp² atoms

- (b) It contains 6π electrons
- (c) 4 nitrogen contain localised lone pairs
- (d) 2 nitrogen contain delocalised lone pairs.
- 5. How many π electrons present in the given molecule.

6. How many atoms are sp² hybridised

Among these canonical structures, the correct order of stability is

(a)
$$I > II > III$$

(b)
$$II > II > I$$

(c)
$$I > III > II$$

(d)
$$II > I > III$$

8. Arrange the following which is most stabilised cation?

9. (I)
$$CH_3 - O - CH = CH - CH = CH_2$$

(II)
$$CH_3 - O - \overset{\oplus}{C}H - CH = CH - \overset{\Theta}{C}H_2$$

(III)
$$CH_3 - \overset{\oplus}{O} = CH - CH = CH - \overset{\Theta}{C}H_2$$

Among these three canonical structures (through more are possible) what would be their relative contribution in the hybrid

(a)
$$I > II > III$$

(b)
$$III > II > I$$

(c)
$$I > III > II$$

(d)
$$III > I > II$$

10.
$$\bigcap_{N} \longleftrightarrow \bigcap_{\Theta} \longleftrightarrow \bigcap_{N} \longleftrightarrow \bigcap_{W} \longleftrightarrow \bigvee_{N} \bigoplus_{V} \longleftrightarrow \bigvee_{N} \longleftrightarrow \bigvee_{N} \bigoplus_{W} \longleftrightarrow \bigvee_{N} \bigvee_{N} \longleftrightarrow \bigvee_{N} \bigvee_{N} \longleftrightarrow \bigvee_{N} \bigvee_{N} \longleftrightarrow \bigvee_{N} \bigvee_{$$

Among these canonical structures of pyridinie, the correct order of stability is

(a)
$$(I = V) > (II = IV) > III$$

(b)
$$(II = IV) > (I = V) > III$$

(c)
$$(I = V) > III > (II = IV)$$

11. In which of the following pairs, first species is more stable than second?

(a)
$$CH_3 - CH_2O$$
 or CH_3CO

O O O O O O (b)
$$CH_3CCHCH_2CH$$
 or CH_3CCHCH_3

(e) CH₃ CHCH₂CCH₃ or CH₃CH₂CHCCH₃

$$(d) \bigvee_{O}^{O} \text{ or } \bigvee_{N^{-}}^{O}$$

12. Which of the following has longest C – O bond:

$$(d) \bigcup_{CH_2}^{O}$$

13. Which of the following carbocation will be more stable?

(d)
$$H_3C - C$$
 CH_3

14. Which of the following is incorrect relation between given pairs?

(a)
$$NH_2$$
 NH_2 = Resonance

(c)
$$\stackrel{S}{\longleftarrow}_{H}$$
 $\stackrel{S^{-}}{\longleftarrow}_{H}$ = Resonance

(d)
$$O_{+}$$
 O_{+} O_{+} O_{+} O_{-} O_{-}

15. Which of the following pair represents non identical pair?

(a)
$$CH_3$$
— $C = N$ — O and CH_3 — C = N — O

(a)
$$CH_3 - C = N - O$$
 and $CH_3 - C = N - O$ (b) $CH_2 = N + O$ and $CH_2 - N + O$ O^{\ominus} and $CH_2 - N + O$

(c)
$$CH_2 = N = N$$
 and $CH_2 - N = N$

(d)
$$C_6H_5$$
— C
 $\overset{\oplus}{\underset{NH_3}{\oplus}}$ and C_6H_5 — C
 $\overset{\oplus}{\underset{NH_2}{\bigcirc}}$

16. The correct stability order of the following resonance structures is:

$$H_2C-N = N$$

$$H_2 \stackrel{\ominus}{C} \stackrel{\oplus}{--} N = N$$

$$H_2\overset{\ominus}{C}-N=\overset{\oplus}{N}$$

(a)
$$I > II > IV > III$$
 (b) $I > III > II > IV$ (c) $II > I > III > IV$ (d) $III . I > IV > II$

17. Which of the following cations is most stable

(a)
$$H_3C$$
 H_3C H_3C

(b)
$$_{\text{H}_3\text{C}} \stackrel{\text{C}}{\stackrel{\oplus}{\longrightarrow}} \stackrel{\text{CH}_3}{\stackrel{}{\longrightarrow}}$$

ANSWER KEY

- 1. c
- 2. b 9. c
- 3. c
- 4. a,d

11. d

5.8

12. b

- 6.9 13. a
- 7. c 14. d

- 8. c 15. d
- 16. b
- 17. c

10. a